

## EXPERIMENTAL INVESTIGATION OF DRY STORAGE CASK HEAT TRANSFER

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### ABSTRACT

Spent nuclear fuel generated at nuclear power plants must be safely stored during interim storage periods. A dry storage cask to safely store the spent nuclear fuel should be able to adequately emit the decay heat from the spent nuclear fuel. Therefore, heat removal tests using a half scale dry storage cask have been performed to estimate the heat transfer characteristics of a dry storage cask under normal. The heat transfer rate to an ambient atmosphere by convective air through a passive heat removal system reached 83%. Accordingly, the passive heat removal system is designed well and works adequately.

### INTRODUCTION

The management of spent nuclear fuel generated at nuclear power plants has become a major policy issue due to continued delays in obtaining a safe, permanent disposal facility. Most nuclear power plants store their spent nuclear fuel in wet storage pools. However, after decades of use, most storage pools have reached maximum capacity. For the nuclear industry, finding sufficient capacity for storage of spent nuclear fuel is essential if the nuclear power plants are to be allowed to continue operation.

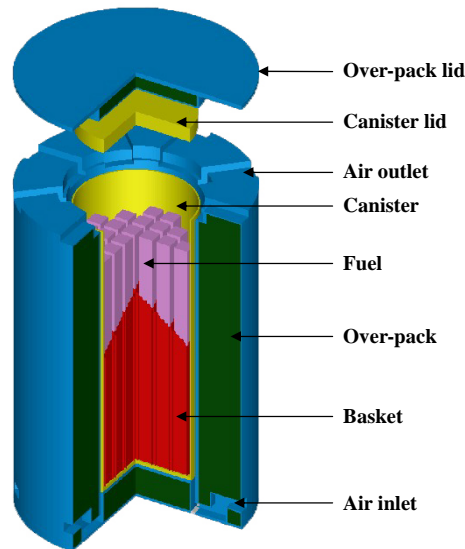
Dry storage casks are one possible solution for solving the interim storage problem. The dry storage cask system consists of three separate components: an over-pack, a canister, and a transfer cask. The spent nuclear fuel assemblies are loaded and sealed inside the canister. Then, the canister is transferred using the transfer cask into a cylindrical over-pack for on-site dry storage. The canister may be removed from the over-pack at any time and transferred into the transport cask for movement to a permanent disposal facility.

Figure 1 shows the schematic of the dry storage cask. The structural casing of the over-pack is made of carbon steel, and the inner cavity of the casing is filled with concrete, which acts as a radiation shield. The dry storage cask must ensure that the temperatures of the spent nuclear fuel assemblies are maintained within the allowable values for normal, off-normal, and accident conditions. Therefore, the dry storage cask must be designed including heat removal capabilities with an appropriate reliability[1]. However, the thermal conductivity of concrete is not good and the allowable temperature of concrete is lower than that of steel. Therefore, a passive heat removal system was designed so that the temperatures of the fuel assembly cladding material and

the dry storage cask components remain within the allowable limits. The passive heat removal system consists of eight inlets and eight outlets [2].

The outer diameter of the dry storage cask is 3,550 mm and its overall height is 5,885 mm. It weighs approximately 135 tons. The dry storage cask accommodates 24 PWR spent fuel assemblies with a burn-up of 55,000 MWD/MTU and a cooling time of 7 years. The decay heat from the 24 PWR spent fuel assemblies is 25.2 kW.

This paper discusses the experimental approach used to evaluate the heat transfer characteristics of the dry storage cask.



**Figure 1. Schematic of the Dry Storage Cask.**

## **HEAT REMOVAL TEST**

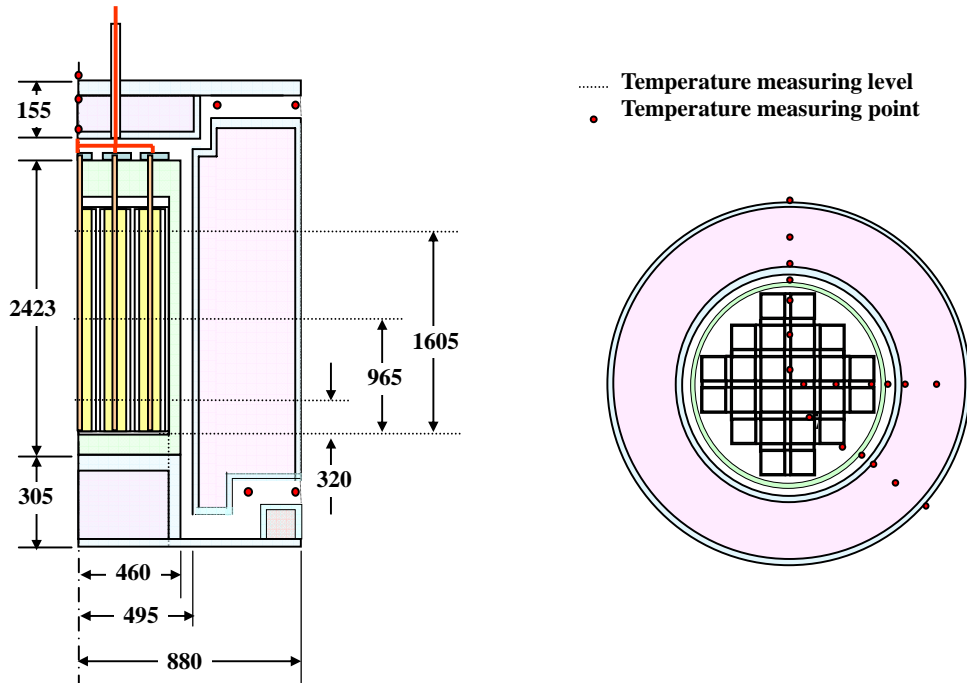
### **Description of the Test Model**

The test model is a one-half scale model of the dry storage cask. Figure 2 shows a cross section of the thermal test model. During actual storage, the lid of the canister is welded to the body of the canister to maintain the confinement. However, during the thermal tests, the lid of the canister was only bolted to the body of the canister to allow the opening of the lid after the test. The lid of the canister had 24 holes for electric heaters and 24 holes for thermocouples. The electric heaters, which were used to simulate 24 PWR spent fuel assemblies, were accommodated within the baskets and fixed onto the top of the lid of the canister with a swage lock.

### **Heat Transfer Mode and Measurement System**

Heat is generated by the spent nuclear fuel assemblies within the canister and transferred to the surface of the canister via conduction, convection, and radiation. This heat is then transferred from the surface of the canister to the inner surface of the over-pack through convection and radiation. The over-pack is designed to dissipate the heat from the canister through a passive heat removal system. This mechanism involves a natural convective air flow through the annular area between the canister and the inner surface of the over-pack. Therefore, heat transfer from the over-pack to the ambient atmosphere is accomplished through two mechanisms: the heat, which

is conducted through the over-pack body, is dissipated from the exterior surface of the over-pack to the ambient atmosphere by convection and radiation, and the air, which is heated in the annular area, is vented to the ambient atmosphere through the outlets of the passive heat removal system.



**Fig. 2. Cross Section of the Thermal Test Model.**

The heat transfer from the exterior surface of the over-pack to the ambient atmosphere is [3,4]:

$$q_s = hA(T_s - T_a) + \sigma \varepsilon A(T_s^4 - T_a^4),$$

where  $q_s$  is the heat flow rate from the exterior surface of the over-pack to the ambient atmosphere,  $h$  is the natural convective heat transfer coefficient,  $A$  is the surface area,  $T_s$  is the temperature at the surface,  $T_a$  is the ambient temperature,  $\sigma$  is the Stefan-Boltzmann constant, and  $\varepsilon$  is the emissivity.

Heat transfer to the ambient atmosphere through the passive heat removal system can be related to the inlet and outlet fluid temperatures through an energy balance [5]:

$$q_A = \dot{m}C_p\Delta T,$$

where  $q_A$  is the heat transfer into the air,  $\dot{m}$  is the mass flow rate,  $C_p$  is the specific heat of the air, and  $\Delta T$  is the differential air temperature from the inlet to the outlet.

Therefore, it is important to estimate the energy balance in order to evaluate the heat transfer characteristics of the dry storage cask. Accordingly, two measurement systems were used in the heat removal test. One was the temperature data acquisition system, which consists of a thermocouple scanner, a signal conditioner, an A/D converter, and a P/C. The other was the velocity data acquisition system, which consists of an anemometer scanner, a data logger, an A/D converter, and a P/C.

**Heat Removal Test**

As shown in Fig. 3, the heat removal tests were carried out in a structure with dimensions of 5.0 m x 5.0 m x 5.0 m under normal conditions [6]. The structure was made from sandwich panels to avoid wind and rain, as well as to decrease the influence of ambient temperature fluctuations. Also, the roof of the house had two exhausts in order to prevent a stratification boundary at the outlet level of the dry storage cask during the heat removal tests.

During the heat removal test, the test model was located in the center of the house, cold air entered the house through six louvers at ground level, and the heated air was vented through the roof. The total heat power from the 24 electric heaters was 4.5 kW.

Table 1 shows the measurement parameters and Fig. 2 shows the measuring points of the temperature in the test model. A total of 105 thermocouples were installed: 79 to measure and monitor the temperature of the test model and 26 to measure and monitor the ambient temperature of the thermal test structure.



Fig. 3. Test Model and Thermal Test Structure.

There were two types of sensor used to measure the air velocity at the inlet and outlet. Hot wire anemometers were used to measure the air velocity at the inlet duct. As the temperature of the exhaust from the outlet duct is very high, the air velocity at the outlet duct was measured with a vane anemometer.

Table 1. Measurement Parameters

Item	Sensor		Subject and Number
Temperature	Thermocouple		Inside of the Over-pack: 18 Surface of the Over-pack: 9 Surface of the Canister: 12 Basket: 24 Inlet: 6 Outlet: 6 Ambient: 26
Air Velocity	Anemometer	Hot Wire	Inlet: 3
		Vane	Outlet: 3

## TEST RESULTS AND DISCUSSION

### 1<sup>ST</sup> TEST RESULT

Table 2 lists the maximum temperatures measured under normal conditions. Thermal equilibrium of the test model was reached after about 120 hours, and that state was maintained for a period of 2 days. The average ambient temperature in the thermal test house was maintained at 21°C approximately during the thermal test. A flow did not occur inside the canister because the gap between the lid of the canister and the basket was quite small at 4 mm approximately. In the basket, therefore, the temperature of the upper and the middle part appeared similarly. In the 135° direction, the temperature of the canister surface and the over-pack was higher than that in the other direction. It showed that the basket was contacted with the canister body.

Table 2. Summary of the Heat Removal Test Results

Location		Maximum temperatures (°C)					
		Basket	Canister	Over-pack		Outlet	Outlet
				Inside	Outside		
0°	Upper	284	110	45	28		70
	Middle	285	106	38	26		
	Lower	212	84	31	24	21	
90°	Upper	283	120	47	28		65
	Middle	286	108	40	26		
	Lower	214	85	32	24	21	
135°	Upper	263	126	47	29		63
	Middle	263	114	40	27		
	Lower	171	87	32	24	21	

### 2<sup>ND</sup> TEST RESULT

Prior to the 2<sup>nd</sup> heat removal test, the test model was modified as follows; the gap between the lid of the canister and the basket had been expanded from 4 mm to 67.5 mm, the cavity between the over-pack and the canister as well as that between the basket and the canister had been adjusted so as to maintain the same distance.

Table 3 lists the maximum temperatures measured from the 2<sup>nd</sup> heat removal test. The thermal equilibrium of the test model was reached after approximately 120 hours, as in the 2<sup>nd</sup> heat removal test, and that state was also maintained for two days. The average ambient temperature in the house was maintained at approximately 27 °C during the test. Figure 4 shows the temperature distribution of the dry storage cask under normal conditions.

The heat transferred from the dry storage cask to the ambient atmosphere was accomplished using the air in the passive heat removal system and heat transfer on the dry storage cask surface. In order to evaluate the energy balance, the heat transfer rate into the air and the heat transfer rate on the dry storage cask surface were calculated using the temperature and velocity data at the inlet and outlet. The difference of the temperature between the inlet and the outlet was

considerably large. The average temperature at the inlet and outlet was measured at 27 °C and 73 °C, respectively. The average velocity at the inlet and outlet was measured at 0.49 m/s and 0.72 m/s, respectively. Therefore, the mass flow rate of the air was calculated to be 0.0104 kg/s. Accordingly, the heat transfer rate to the ambient atmosphere by the air was estimated as 83% of the heat transferred from the dry storage cask to the environment. This shows that the passive heat removal system was designed well and worked adequately.

Table 3. Test Results for the 2<sup>nd</sup> Heat Removal test

Location		Maximum temperatures (°C)					
		Basket	Canister	Over-pack		Inlet	Outlet
				Inside	Outside		
0°	Upper	259	116	53	36		70
	Middle	240	105	46	34		
	Lower	177	-	39	32	27	
90°	Upper	259	121	54	36		76
	Middle	239	110	47	34		
	Lower	176	88	40	32	27	

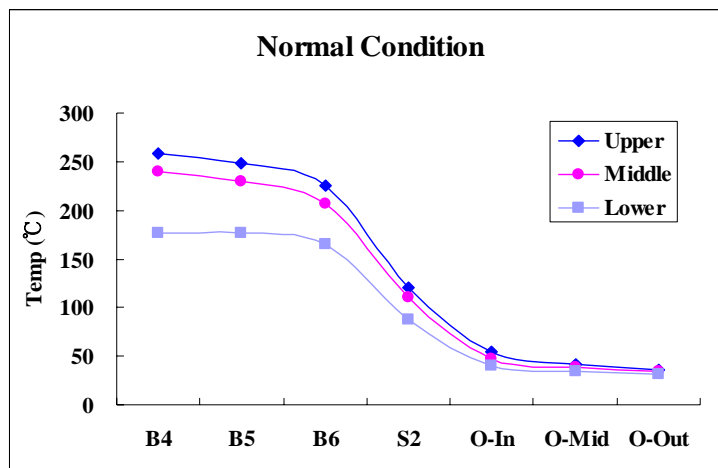


Fig. 4. Temperature Distribution of the Dry Storage Cask under Normal Conditions.

## CONCLUSIONS

A heat removal test was carried out to evaluate the heat transfer characteristics of a dry storage cask. The main results are as follows:

- (i) The heat transfer rate to the ambient atmosphere by convective air through a passive heat removal system reached 83%. Accordingly, the passive heat removal system was well designed and worked adequately.
- (ii) To prevent a local temperature rise, the cavity between the canister and the over-pack as well as that between the basket and the canister has to be designed so as to maintain the same distance.
- (iii) The gap between the lid of the canister and the basket must have an appropriate space to ensure that the flow can happen inside the canister.

## **ACKNOWLEDGMENTS**

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